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Stereoselectivity investigation on glycosylation of oxazolidinone protected 2-amino-2-deoxy-D-glucose donors based on pre-activation protocol

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ABSTRACT

Diverse 2,3-oxazolidinone protected 2-amino-2-deoxy-D-glucose thioglycosides were prepared and studied as glycosyl donors at low temperature by BSM/Tf2O pre-activation protocol before the addition of glycosyl acceptors. The stereochemistry outcomes of a series of glycosylations were investigated. Different stereoselectivities of the coupling reactions were obtained, arising from the different protecting groups in the oxazolidinone donors. 4,6-Di-O-benzyl-N-benzyl-oxazolidinone protected thioglycoside donor 1c underwent glycosylation with general β -anomeric selectivity and the stereoselectivity could be also affected by glycosylation conditions.

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1. Introduction

Stereoselective glycosylation for 2-amino-2-deoxy sugars is critical to the assembly of numerous aminosugar-containing oligosaccharides and glycoconjugates with biologically important roles.[1,2](#page-9-0) In particular, formation of 1,2-cis glycosidic bond, requiring the use of a non-participating group such as an azido group at $C-2$,^{[3](#page-9-0)} is still a challenge to synthetic chemists.^{[4](#page-9-0)} In 2001, Kerns and coworkers reported 2,3-trans-oxazolidinone as an effective protective group possessing high α -selectivity toward glycosylations.^{[5a](#page-9-0)} Unfortunately, this glycosyl donor has several drawbacks such as the undesired glycosylation and sulfenylation of the nitrogen atom. Later, Kerns and co-workers^{[5b,c](#page-9-0)} and Oscarson and co-workers^{[6](#page-9-0)} modified the oxazolidinone group to its N-acetyl analog, leading to significant reduction of α -selectivity, and β -selectivity was observed in some cases. Recently, Ito and co-workers disclosed N-benzyl-2,3-trans-oxazolidinone as a highly α -directing donor,^{[7](#page-9-0)} but the coupling yields are generally not high and the origin of selectivity is not clear. To tackle these problems, a new efficient strategy for both α - and β -stereoselective glycosylations of glucosamine donors based on pre-activation protocol was developed by us very recently. 8 By virtue of the pre-activation strategy, $9,10$ that is, a glycosyl donor was pre-activated in the absence of an acceptor, the known 4,6-di-O-acetyl-N-acetyl-oxazolidinone protected

donor $1a$ afforded either excellent β -stereoselectivity or excellent a-stereoselectivity toward glycosylations simply by means of the addition of hindered base 2,4,6-tri-tert-butylpyrimidine (TTBP)¹¹ or the absence of base (Fig. 1).

Our initial success suggested that pre-activation protocol could greatly influence the stereochemistry outcomes of glycosylations. Meanwhile, under the conditions of non-preactivation, change and uncertainty in stereochemistry of glycosylations by the use of oxazolidinone and its derivatives as glycosyl donors reported by different research groups^{[5–7](#page-9-0)} sparked our curiosity. To explore the scope of pre-activation protocol and investigate the influence of protecting groups at the 4,6-OH and/or 2-NH positions on stereochemistry outcomes toward glycosylations, our attention was paid to stereoselective glycosylations with various oxazolidinone-containing glycosyl donors by pre-activation manner, and our findings are reported here.

Figure 1. Stereoselectivity-controllable glycosylation of donor 1a.

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2. Results and discussions

2.1. Preparation of glycosyl donors 1b–1g

A series of oxazolidinone protected glucosamine thioglycosides 1b–1g were designed and synthesized. In these glycosyl donors, the 4,6-OH and 2-NH functional groups were protected by two kinds of groups: the electron-donating benzyl group (or benzylidene group) and the electron-withdrawing acetyl group. Among them, 1b is known and was prepared according to the literature. $\frac{5}{3}$

$$
\begin{array}{c}\nACO & O \\
ACO & H \\
\hline\nO & H\n\end{array}
$$
STol

The synthesis of N-benzyl-oxazolidinone protected glycosyl donors 1c, 1f and 1g, N-acetyl-oxazolidinone protected glycosyl donor 1e and N-unsubstituted-oxazolidinone protected glycosyl donor 1d all started from the same glucosamine derivative 2^{12} 2^{12} 2^{12} (Scheme 1). Removal of the acetyl groups in 2 with 30% MeONa afforded the intermediate 3 in quantitative yield. Compound 3 was treated with sodium hydride (NaH) and benzyl bromide (BnBr) to provide 4,6 di-O-benzyl-N-benzyl-2N,3O-oxazolidinone protected donor 1c and 4,6-di-O-benzyl-2N,3O-oxazolidinone protected donor 1d, respectively, just by using different amount of the benzylation reagents (6.0 equiv of NaH and 4.5 equiv of BnBr for 1c; 4.5 equiv of NaH and 2.5 equiv of BnBr for 1d). The reaction outcomes can be modulated because the O-benzylation occurs ahead of the N-benzylation. The N-acetyl donor 1e was easily prepared via acetylation of donor 1d. On the other hand, compound 3 was subjected to 4,6-O-benzylidene protection to afford 4, which was treated again with NaH and BnBr, yielding the N-benzyl donor 1f in good isolated yield. Subsequently, 1f was smoothly converted to 1g via selective deprotection followed by re-acetylation.

2.2. Glycosylations of various 2,3-oxazolidinone protected glycosyl donors with unalterable acceptor 6a

With glycosyl donors 1b-1g in hand, we firstly investigated the influence of protecting groups at the 4,6-OH and/or 2-NH positions on stereochemistry outcomes toward glycosylations. A series of couplings of the unalterable acceptor 6a with diverse donors 1b–1g by pre-activation protocol were checked ([Table 1\)](#page-2-0). Identical to our previous work,^{[8](#page-9-0)} a combination of benzenesulfinyl morpholine $(BSM)^{13}$ $(BSM)^{13}$ $(BSM)^{13}$ and triflic anhydride (Tf₂O) acted as the promoter system. Since the use of hindered base is not necessary in our routine preactivation protocol, $9,13$ the glycosylation reactions were conducted in the absence of hindered base TTBP. Each donor was pre-activated by BSM/Tf₂O at low temperature in anhydrous $CH₂Cl₂$, the acceptor 6a was then added to the reaction mixture after the complete consumption of donor by TLC monitoring. As shown in [Table 1,](#page-2-0) glycosyl donors 1b, 1c, 1e and 1g successfully afforded the coupling products in good isolated yields, 14 but interestingly, the stereoselectivities varied with the glycosyl donors equipped with different protecting groups. Among them, the tri-benzyl substituted donor 1c exhibited complete β -selectivity (entry 2). Compared with 4,6-di-O-acetyl-N-acetyl-oxazolidinone protected donor 1a reported in our previous work, glycosyl donor 1c afforded the completely opposite stereochemistry outcomes under the same glycosylation conditions. It showed that the stereoselectivities of glycosylations by pre-activation protocol can be modulated by suitably protected glycosyl donors.

The different stereoselectivities might arise from the variety of intermediates. After being activated, the oxazolidinone protected donors turned into the oxacarbenium ion, which could be trapped by triflate anion to give either α -triflate or β -triflate.¹⁵ Thus, under pre-activation conditions, there are three types of intermediate: oxacarbenium ion, α -triflate and β -triflate. Oxacarbenium ion intermediate may participate in an S_N 1-like process with the acceptor, while triflate intermediate may participate in an S_N2 -like process. The electronic property of protecting groups may influence the major form of intermediate during the glycosylation reactions. Pre-activation of 4,6-O-benzyl protected donors 1c and 1e (entries 2 and 4, [Table 1](#page-2-0)) produced oxacarbenium ions, which could be the major form existed instead of the glycosyl triflate presumably due to the stabilization effect of electron-donating benzyl groups. Attacked by acceptor $6a$ via an S_N1 -like process, the disaccharides 7c and 7e were gained. The only discrimination between 1c and 1e is the protecting group at 2-N-position. N-Benzyl group affords steric hindrance to α -face attack by hindered nucleophiles, and results in excellent β -selectivity. This steric hindrance is removed in the case of N-acetyl substituent, so there is no selectivity for donor 1e. Pre-activation of 4,6-0-acetyl protected donors 1b and 1g (entries 1 and 6, [Table 1](#page-2-0)) could afford glycosyl triflates by contrast. The equilibrium might favor the formation of covalent anomeric triflates in this case because the oxacarbenium ion could be very unstable and be trapped easily by triflate anion, arising from the electron-withdrawing effect of acetyl groups. Both a-triflate and β -triflate may exist, so the coupling products are α/β -anomeric mixtures.

2.3. Stereoselective couplings of donor 1c with various glycosyl acceptors

Based on the excellent β -stereoselectivity of coupling between donor 1c and acceptor 6a, we then focused on performing

Scheme 1. Preparation of donors 1c-1g. Reagents and conditions: (a) MeONa, MeOH, 100%; (b) 4.5 equiv BnBr, 6.0 equiv NaH, DMF, 80%; (c) 2.5 equiv BnBr, 4.5 equiv NaH, DMF, 65%; (d) Ac₂O, Py, 95% for **1e**, 90% for **1g**; (e) PhCH(OMe)₂, CSA, CH₃CN, 85%; (f) 1.5 equiv BnBr, 2.5 equiv NaH, DMF, 90%; (g) TsOH, MeOH, 85%.

Table 1

The couplings of various 2,3-oxazolidinone protected glycosyl donors with unalterable acceptor 6a

 $^{\rm a}$ Anomeric ratio was determined by the integration of ¹H NMR spectrum of the anomeric mixture.

glycosylations of 1c with a series of acceptors by pre-activation manner to illustrate the generality of the stereoselectivity, and the results are listed in [Table 2.](#page-3-0) As shown, donor 1c was β -selective toward glycosylations with a wide range of glycosyl acceptors except for acceptor 6m (entry 13) and the coupling yields were generally high. And a clear trend that the stereochemistry outcomes of the glycosylations correlated to acceptors was observed. Acceptors 6a–6e equipped with 4,6-O-benzylidene protecting group displayed excellent β -selectivity (entries 1–5), and more reactive acceptors 6i and 6j with primary hydroxyl group also led to good β -selectivity (entries 9 and 10). In contrast to the recently reported similar donor N-benzyl-2,3-trans oxazolidinone derivative, λ our donor **1c** provided opposite stereoselectivity under pre-activated conditions. It is again verified that pre-activation protocol can influence the stereoselectivity of glycosylations.

According to the results displayed in [Table 2](#page-3-0), the glycosylation might be an S_N 1-like process via the intermediate of oxacarbenium ion. The benzyl group on nitrogen atom affords steric hindrance to a-face attack by hindered nucleophiles. Thus the glycosylation r eaction is β -selective, more hindered acceptors equipped with the 4,6-O-benzylidene protecting group afford better stereoselectivity. In comparison with donor ${\bf 1a}^8$ ${\bf 1a}^8$ the opposite stereoselectivity of donor 1c may arise from the different intermediates produced after pre-activation, which could be influenced by the electronic property of protecting groups at the 4,6-OH and 2-NH positions.

2.4. Stereochemistry under different glycosylation conditions

Next we turned our attention to the influence on stereochemistry outcomes arising from different glycosylation conditions such as solvent, temperature and promoter system. The representative reaction was performed as exemplified by the coupling of donor 1c and acceptor 6a ([Table 3\)](#page-3-0). In contrast to our published work,^{[8](#page-9-0)} addition of the hindered base TTBP to the reaction system gave rise to the same stereoselectivity in a little higher yield (entry 1). Under pre-activation conditions, using benzenesulfinyl piperidine/triflic anhydride $(BSP/TF_2O)^{16}$ $(BSP/TF_2O)^{16}$ $(BSP/TF_2O)^{16}$ or diphenyl sulfoxide/triflic anhydride $(Ph_2SO/Tr_2O)^{17}$ $(Ph_2SO/Tr_2O)^{17}$ $(Ph_2SO/Tr_2O)^{17}$ instead of our BSM/Tf₂O promoter system, the coupling afforded almost the same results (entry 2). However, activation of thioglycoside 1c by employing N-(phenylthio)- ε -caprolactam/Tf₂O¹⁸ or NIS/AgOTf^{[19](#page-9-0)} led to non-stereoselective glycosylation, providing both α - and β -linked products (entries 3–6). The quotient of α -anomer increased along with rise of temperature (entry 3 vs entry 4). Change of solvent from CH_2Cl_2 to toluene also led to the increase of a-anomer (entry 3 vs entry 5). It was exhibited that stereochemistry outcomes of glycosylations can be modulated by reaction conditions such as promoter, temperature, solvent and activation manner. The rise of temperature and the use of non-polar solvent resulted in the increase of the thermodynamic-controlled product a-anomer.

Table 2

Glycosylations of donor 1c with a series of acceptors by BSM/Tf_2O pre-activation

^a Anomeric ratio was determined by the integration of ¹H NMR spectrum of the anomeric mixture.

2.5. Deprotection of disaccharide 8d

The formed disaccharides can be fully deprotected. For example, as shown in [Scheme 2](#page-4-0), disaccharide 8d was deprotected under basic conditions (t -BuOK/DMSO).²⁰ The ring-fused oxazolidinone was thus opened in several minutes in excellent yield affording 9, followed by the concomitant removal of the O-, N-benzyl groups and O-benzylidene group over hydrogenolysis to obtain disaccharide 10.

3. Conclusion

We have demonstrated that diverse 2,3-oxazolidinone protected 2-deoxy-2-amino-D-glucose thioglycosides show different even opposite stereoselectivities in glycosylations conducted by the BSM/Tf₂O pre-activation protocol at low temperature in dichloromethane. The stereochemistry outcomes of coupling reactions are strongly influenced by the protecting groups in oxazolidinone protected glycosyl donors and the activation manner of donors. In contrast to the tri-acetyl protected donor 1a reported previously, 4,6-di-O-benzyl-N-benzyl-oxazolidinone protected thioglycoside donor $1c$ affords moderate to excellent β -selectivity toward glycosylations. Moreover, in the case of donor 1c, the stereoselectivity

^a Anomeric ratio was determined by the integration of ¹H NMR spectrum of the anomeric mixture.

Scheme 2. Deprotection of disaccharide 8d. Reagents and conditions: (a) t -BuOK, DMSO, 90%; (b) H_2 , Pd/C, 95%.

cannot be affected by the addition of TTBP, but can be modulated by changing reaction conditions such as solvent, temperature and promoter system. In summary, based on our BSM/Tf₂O pre-activation strategy, using 2,3-oxazolidinone protected thioglycoside donors, either α - or β -glycosidic linkage can be formed. The stereoselectivity can be greatly affected by modulating protective groups of the glycosyl donors. The mechanistic understandings await further investigation.

4. Experimental

4.1. General procedures

All chemicals were purchased as reagent grade and used without further purification, unless otherwise noted. Dichloromethane $(CH₂Cl₂)$, pyridine, toluene and acetonitrile (CH₃CN) were distilled over calcium hydride (CaH2). Methanol was distilled from magnesium. DMF was stirred with $CaH₂$ and distilled under reduced pressure. All reactions were carried out under anhydrous conditions with freshly distilled solvents, unless otherwise noted. Reactions were monitored by analytical thin-layer chromatography on silica gel 60 F254 precoated on aluminum plates (E. Merck). Spots were detected under UV (254 nm) and/or by staining with acidic ceric ammonium molybdate. Solvents were evaporated under reduced pressure and below 40 \degree C (bath). Organic solutions of crude products were dried over anhydrous $Na₂SO₄$. Column chromatography was performed on silica gel (200–300 mesh). $^1\mathrm{H}$ NMR spectra were recorded on a JEOL AL-300, Varian INOVA-500, or Advance DRX Bruker-500 spectrometers at 25 °C. Chemical shifts (in ppm) were referenced to tetramethylsilane (δ =0 ppm) in deuterated chloroform. 13 C NMR spectra were obtained by using the same NMR spectrometers and were calibrated with CDCl₃ (δ = 77.00 ppm). Mass spectra were recorded using a PE SCLEX QSTAR spectrometer. Elemental analysis data were recorded on a Vario EL-III elemental analyzer.

4.2. Preparation of oxazolidinone protected glycosyl donors

4.2.1. p-Tolyl N-2,2,2-trichloroethyloxycarbonyl-2-amino-2-deoxy-1-thio- β -*p*-glucopyranoside (3)

Trichloroethyl carbamate protected glucosamine 2 (2.00 g, 3.41 mmol, 1 equiv) in 100 mL methanol containing a catalytic amount of NaOMe (30% in MeOH) under pH 8–9 was stirred for 1.5 h. The reaction mixture was neutralized with cation exchange resin (H^+) , filtered and concentrated to dryness under reduced pressure to give compound 3 (1.50 g, 95%) as a white solid. R_f =0.7 (ethyl acetate/methanol, 10:1). Compound 3 was used for the next reaction directly.

4.2.2. p-Tolyl N-benzyl-2-amino-2,3-N,O-carbonyl-4,6-O-dibenzyl-2-deoxy-1-thio- β -D-glucopyranoside (1c)

Compound 3 (1.50 g, 3.26 mmol, 1 equiv) was dissolved in dry DMF (40 mL) and stirred at -15 °C, and NaH (0.78 g, 19.57 mmol, 6 equiv, 60% in mineral oil) was added, followed by the addition of benzyl bromide (1.75 mL, 14.67 mmol, 4.5 equiv). After stirring for 30 min on the ice-water bath, the reaction mixture was warmed to

room temperature and stirred for 4 h. The mixture was directly concentrated under reduced pressure, and the residue was purified by column chromatography (petroleum ether/ethyl acetate, 12:1) to give $1c$ (1.51 g, 80%) as a white solid: $^1\mathrm{H}$ NMR (500 MHz, CDCl₃) δ 7.24–7.41 (m, 17H), 7.00 (d, 2H, J=8.0 Hz), 4.88 (d, 1H, J=11.5 Hz), 4.78 (d, 1H, J=15.0 Hz), 4.74 (d, 1H, J=15.0 Hz), 4.69 (d, 1H, J=9.5 Hz, H-1), 4.55 (d, 1H, J=11.5 Hz), 4.54 (d, 1H, J=11.5 Hz), 4.48 (d, 1H, $J=12.0$ Hz), 4.16 (dd, 1H, $J=10.0$, 11.0 Hz), 3.86 (dd, 1H, $J=8.5$, 9.5 Hz), 3.75 (dd, 1H, J=2.0, 11.0 Hz), 3.69 (dd, 1H, J=5.0, 11.0 Hz), 3.54-3.57 $(m, 1H)$, 3.44 (dd, 1H, J=9.5, 11.0 Hz), 2.29 (s, 3H); ¹³C NMR (125 MHz, CDCl3) d 159.27, 138.61, 138.06, 137.32, 136.36, 132.93, 129.80, 128.63, 128.48, 128.39, 128.33, 128.16, 128.00, 127.93, 127.70, 127.62, 127.57, 86.64, 83.45, 79.91, 73.74, 73.38, 73.16, 68.53, 60.30, 47.46, 21.10; MS (FAB) 581 [M]⁺; Anal. Calcd for C₃₅H₃₅NO₅S: C, 72.26; H, 6.06; N, 2.41. Found: C, 72.37; H, 6.15; N, 2.31.

4.2.3. p-Tolyl 2-amino-2,3-N,O-carbonyl-4,6-O-dibenzyl-2-deoxy-1-thio- β -*p*-glucopyranoside (1**d**)

Compound 3 (1.50 g, 3.26 mmol, 1 equiv) was dissolved in dry DMF (30 mL) and stirred at -15 °C, and NaH (0.59 g, 14.67 mmol, 4.5 equiv, 60% in mineral oil) was added, followed by the addition of benzyl bromide (0.97 mL, 8.15 mmol, 2.5 equiv). After stirring for 30 min on the ice-water bath, the reaction mixture was warmed to room temperature and stirred for 2 h. The mixture was directly concentrated under reduced pressure, and the residue was purified by column chromatography (petroleum ether/ethyl acetate, 5:1) to give the major product ${\bf 1d}$ (1.04 g, 65%) as a white solid: ¹H NMR (500 MHz, CDCl3) d 7.28–7.41 (m, 10H), 7.15–7.17 (m, 2H), 7.07–7.09 (m, 2H), 4.91 (d, 1H, J=11.5 Hz), 4.78 (s, 2H), 4.72 (d, 1H, J=9.5 Hz, H-1), 4.61 (d, 1H, J=11.0 Hz), 4.18 (dd, 1H, J=10.0, 11.0 Hz), 3.80-3.85 (m, 2H), 3.69–3.74 (m, 1H), 3.45–3.48 (m, 1H), 3.36 (dd, 1H, J=9.5, 11.5 Hz), 2.32 (s, 3H); ¹³C NMR (125 MHz, CDCl₃) δ 159.21, 139.03, 137.20, 136.21, 133.06, 129.94, 128.70, 128.50, 128.15, 128.05, 127.65, 86.70, 83.19, 80.19, 73.49, 73.22, 61.60, 60.43, 47.52, 21.12; HRMS (ESI) calcd for $C_{28}H_{30}NO_5S$ [M+H]⁺: 492.1839, found: 492.1826.

4.2.4. p-Tolyl N-acetyl-2-amino-2,3-N,O-carbonyl-4,6-O-dibenzyl-2-deoxy-1-thio- β -D-glucopyranoside (1e)

To a stirred solution of the compound 1d (0.50 g, 1.02 mmol, 1 equiv) in pyridine (10 mL), was added acetic anhydride (0.29 mL, 3.05 mmol, 3 equiv). The reaction mixture was stirred at room temperature for 2 h, quenched by the addition of saturated aqueous NaHCO₃ (15 mL) and extracted with CH₂Cl₂ ($2\times$ 15 mL). The combined organic layer was dried, filtered and concentrated under reduced pressure. The residue was purified by column chromatography (petroleum ether/ethyl acetate, 5:1) to give 1e (0.51 g, 95%) as a white solid: ¹H NMR (500 MHz, CDCl₃) δ 7.28–7.42 (m, 10H), 7.21-7.23 (m, 2H), 7.07 (d, 2H, J=8.0 Hz), 4.90 (d, 1H, $J=11.5$ Hz), 4.49 (d, 1H, J=15.5 Hz), 4.75 (d, 1H, J=16.0 Hz), 4.66 (d, 1H, J=9.5 Hz, H-1), 4.59 (d, 1H, J=11.5 Hz), 4.30 (dd, 1H, J=2.5, 12.0 Hz), 4.25 (dd, 1H, J=5.0, 12.0 Hz), 4.16 (dd, 1H, J=9.5, 11.0 Hz), 3.74 (t, 1H, J=9.5 Hz), 3.58-3.61 (m, 1H), 3.40 (dd, 1H, J=9.5, 11.5 Hz), 2.33 (s, 3H), 1.96 (s, 3H); ¹³C NMR (125 MHz, CDCl₃) d 170.49, 159.08, 138.91, 136.93, 136.29, 133.24, 129.78, 128.67, 128.51, 128.28, 128.19, 128.14, 127.63, 86.69, 83.37, 77.76, 73.05,

72.96, 62.74, 60.30, 47.49, 21.13, 20.75; HRMS (ESI) calcd for $C_{30}H_{31}NO_6$ SNa [M+Na]⁺: 572.1504, found: 572.1506.

4.2.5. p-Tolyl N-2,2,2-trichloroethyloxycarbonyl-2-amino-4,6-Obenzylidene-2-deoxy-1-thio- β -D-glucopyranoside (4)

Compound 3 (1.50 g, 3.26 mmol, 1 equiv) in CH_3CN (40 mL) containing a catalytic amount of camphorsulfonic acid under PH 2– 3 was stirred, followed by the addition of benzaldehyde dimethyl acetyl (0.59 mL, 3.91 mmol, 1.2 equiv). The mixture was stirred for 5 h, then neutralized with Et_3N and concentrated to dryness under reduced pressure. The crystalline residue was crystallized from EtOAc to give 4 (1.52 g, 85%) as a white solid. The NMR data of 4 was identical to that reported previously.^{[9b](#page-9-0)}

4.2.6. p-Tolyl N-benzyl-2-amino-4,6-O-benzylidene-2,3-N,Ocarbonyl-2-deoxy-1-thio- β - D -glucopyranoside (1f)

Compound 4 (1.00 g, 1.82 mmol, 1 equiv) was dissolved in dry DMF (15 mL) and stirred at -15 °C, and NaH (0.18 g, 4.56 mmol, 2.5 equiv, 60% in mineral oil) was added, followed by the addition of benzyl bromide (0.33 mL, 2.74 mmol, 1.5 equiv). After stirring for 30 min on the ice-water bath, the reaction mixture was warmed to room temperature and stirred for 2 h. The mixture was directly concentrated under reduced pressure, and the residue was purified by column chromatography (petroleum ether/ethyl acetate, 10:1) to give the main product ${\bf 1f}$ (0.89 g, 90%) as a white solid: ¹H NMR (500 MHz, CDCl3) d 7.43–7.26 (m, 10H), 7.11–7.07 (m, 4H), 5.73 (s, 1H), 5.28 (d, J=9.5 Hz, 1H), 4.71-4.66 (m, 2H), 4.59 (d, J=11.5 Hz, 1H), 4.25–4.20 (m, 2H), 3.86 (t, J=10.5 Hz, 1H), 3.74–3.69 (m, 1H), 3.67 (t, $J=10.5$ Hz, 1H), 2.26 (s, 3H); ¹³C NMR (125 MHz, CDCl₃) d 158.6, 137.7, 17.4, 137.0, 131.4, 129.8, 129.1, 128.6, 128.4, 128.2, 127.0, 100.2, 85.8, 77.5, 71.5, 67.4, 61.3, 47.4, 20.6; HRMS (ESI) calcd for $C_{28}H_{28}NO_5S$ [M+H]⁺: 490.1683, found: 490.1682.

4.2.7. p-Tolyl N-benzyl-2-amino-2,3-N,O-carbonyl-2-deoxy-1-thio- β -*D*-glucopyranoside (5)

Compound 1f (0.50 g, 1.02 mmol, 1 equiv) was dissolved in methanol (15 mL) and stirred, then TsOH (0.10 g, 0.51 mmol, 0.5 equiv) was added. The reaction mixture was stirred over night, then neutralized with Et_3N and concentrated to dryness under reduced pressure to give compound 5 (0.35 g, 86%) as a white solid. R_f =0.1 (petroleum ether/ethyl acetate, 10:1). Compound 5 was used for the next reaction directly.

4.2.8. p-Tolyl N-benzyl-2-amino-2,3-N,O-carbonyl-4,6-O-diacetyl-2-deoxy-1-thio- β - D -glucopyranoside (1g)

To a stirred solution of compound 5 (0.35 g, 0.87 mmol, 1 equiv) in pyridine (10 mL), was added acetic anhydride (0.82 mL, 8.71 mmol, 10 equiv). The reaction mixture was stirred at room temperature for 3 h, quenched by the addition of saturated aqueous NaHCO₃ (15 mL) and extracted with CH₂Cl₂ ($2\times$ 15 mL). The combined organic layer was dried, filtered and concentrated under reduced pressure. The residue was purified by column chromatography (petroleum ether/ethyl acetate, 3:1) to give 1g (0.38 g, 90%) as a white solid: ^1H NMR (300 MHz, CDCl₃) δ 7.23–7.41 (m, 7H), 7.09 (d, 2H, J=8.0 Hz), 5.24 (dd, 1H, J=9.0, 10.0 Hz), 4.76 (s, 2H), 4.71 (d, 1H, J=9.6 Hz, H-1), 4.10–4.26 (m, 3H), 3.67–3.73 (m, 1H), 3.50 (dd, 1H, J=9.6, 11.0 Hz), 2.34 (s, 3H), 2.10 (s, 3H), 2.07 (s, 3H); ¹³C NMR (125 MHz, CDCl₃) δ 170.45, 169.18, 158.52, 139.10, 135.98, 133.41, 129.83, 128.72, 128.20, 127.86, 127.73, 86.89, 79.82, 77.17, 67.42, 62.22, 60.19, 47.52, 21.14, 20.71, 20.62; HRMS (ESI) calcd for $C_{25}H_{28}NO_{7}S$ [M+H]⁺: 486.1581, found: 486.1571.

4.3. Glycosylations of donors 1b–1g with acceptor 6a

General procedure: Triflic anhydride (11.8 µL, 0.070 mmol, 1.3 equiv) was added to a stirred solution of 1b–1g (1.2 equiv), BSM (15.9 mg, 0.075 mol, 1.4 equiv) and activated 4 Å molecular sieves (400 mg) in CH_2Cl_2 (5.0 mL) at low temperature under nitrogen atmosphere. The reaction mixture was stirred for 5–20 min, after loss of donor detected by TLC, a solution of the acceptor alcohol **6a** (20.0 mg, 0.054 mmol, 1.0 equiv) in CH_2Cl_2 (0.5 mL) was added dropwise to the pre-activated system. The mixture was stirred and warmed to room temperature slowly, and then quenched by Et_3N (0.1 mL). The precipitate was filtered off and the filtrate was concentrated. The residue was purified by column chromatography on silica gel to give 7b, 7c, 7e and 7g.

4.3.1. Coupling of 1b with 6a to give methyl (2-amino-2,3-N,Ocarbonyl-4,6-diacetyl-2-deoxy- α - and β -p-glucopyranosyl)-(1 \rightarrow 3)-2-O-benzyl-4,6-O-benzylidene-a-D-glucopyranosides

(7 $\mathbf{b}\alpha$) and (7 $\mathbf{b}\beta$)

Donor 1b was pre-activated at -73 °C, and then the reaction mixture was quenched by Et_3N at -60 °C. The crude product was purified by column chromatography (petroleum ether/ethyl acetate, 3:1 to 1.5:1) to give **7b**, yield=87% (α/β =1:2). R_f (α -isomer)=0.25, R_f (β -isomer)=0.20 (petroleum ether/ethyl acetate, 1:1); compound **7b** α : ¹H NMR (500 MHz, CDCl₃) δ 7.36-7.41 (m, 10H), 5.49 (s, 1H), 5.44 (d, 1H, J=2.5 Hz, H-1^t), 5.23 (t, 1H, $J=10.0$ Hz), 4.83 (s, 1H), 4.73 (d, 1H, $J=3.5$ Hz, H-1), 4.59-4.66 (m, 3H), 4.27 (t, 1H, J=9.5 Hz), 4.26 (dd, 1H, J=4.5, 10.5 Hz), 4.12-4.15 (m, 1H), 4.00 (dd, 1H, J=2.0, 12.5 Hz), 3.94 (dd, 1H, J=4.0, 12.5 Hz), 3.83 (dt, 1H, J=5.0, 10.0 Hz), 3.71 (t, 1H, J=10.5 Hz), 3.56-3.63 (m, 3H), 3.41 (s, 3H), 2.06 (s, 6H); ¹³C NMR (125 MHz, CDCl₃) δ 170.63, 169.15, 158.47, 137.35, 136.95, 129.50, 128.61, 128.57, 128.32, 126.09, 101.92, 98.28, 95.29, 82.20, 77.67, 76.08, 74.41, 72.84, 70.04, 69.00, 68.10, 61.90, 61.26, 58.47, 55.40, 20.68 (2C); HRMS (ESI) calcd for $C_{32}H_{37}NO_{13}Na$ [M+Na]⁺: 666.2157, found: 666.2155. Compound **7b** β : ¹H NMR (500 MHz, CDCl₃) δ 7.33–7.47 (m, 10H), 5.52 (s, 1H), 5.30 (s, 1H), 5.24 (dd, 1H, J=9.0, 10.0 Hz), 4.88 (d, 1H, J=8.0 Hz, H-1 \prime), 4.67 (d, 1H, J=11.5 Hz), 4.66 (d, 1H, J=4.0 Hz, H-1), 4.64 (d, 1H, $J=11.5$ Hz), 4.25 (dd, 1H, J=5.0, 10.0 Hz), 4.09–4.19 (m, 3H), 4.04 (dd, 1H, $J=10.5$, 12.0 Hz), 3.79 (dt, 1H, $J=4.5$, 10.0 Hz), 3.71 (t, 1H, $J=10.0$ Hz), 3.62–3.66 (m, 1H), 3.58 (dd, 1H, $J=3.5$, 9.5 Hz), 3.56 (t, 1H, $J=9.5$ Hz), 3.43 (dd, 1H, $J=7.5$, 12.0 Hz), 3.39 (s, 3H), 2.08 (s, 3H), 1.98 (s, 3H); ¹³C NMR (125 MHz, CDCl₃) δ 170.63, 169.15, 158.22, 137.11, 137.03, 129.06, 128.83, 128.69, 128.33, 128.18, 125.92, 101.85, 101.19, 98.27, 79.79, 79.26, 78.95, 76.62, 74.76, 73.53, 68.80, 67.49, 62.40, 62.18, 59.93, 55.41, 20.67, 20.63; MS (FAB) 644 $[M+H]^{+}$; Anal. Calcd for $C_{32}H_{37}NO_{13}$: C, 59.71; H, 5.79; N, 2.18. Found: C, 60.06; H, 5.66; N, 1.99.

4.3.2. Coupling of 1c with 6a to give methyl (N-benzyl-2-amino-2,3-N,O-carbonyl-4,6-dibenzyl-2-deoxy- β -D-glucopyranosyl)-(1 \rightarrow 3)-2-O-benzyl-4,6-O-benzylidene- α -D-glucopyranoside (7c)

Donor 1c was pre-activated at -73 °C. The crude product was purified by column chromatography (petroleum ether/ethyl acetate, 6:1) to give **7c**, yield=85%. R_f =0.40 (petroleum ether/ethyl acetate, 2:1); ¹H NMR (500 MHz, CDCl₃) δ 7.39–7.46 (m, 4H), 7.13– 7.34 (m, 21H), 5.48 (s, 1H), 5.11 (d, 1H, J=8.0 Hz, H-1^t), 4.77 (d, 1H, J=11.5 Hz), 4.69 (d, 1H, J=3.5 Hz, H-1), 4.54 (d, 1H, J=15.0 Hz), 4.49 $(s, 2H)$, 4.45 (d, 1H, J=11.0 Hz), 4.37 (d, 1H, J=14.5 Hz), 4.36 (d, 1H, $J=12.0$ Hz), 4.36 (t, 1H, $J=9.0$ Hz), 4.31 (d, 1H, $J=12.0$ Hz), 4.26 (dd, 1H, $J=4.5$, 10.0 Hz), 3.98 (dd, 1H, $J=10.0$, 12.0 Hz), 3.82 (dd, 1H, $J=4.5$, 10.0 Hz), 3.78 (dd, 1H, $J=8.5$, 9.5 Hz), 3.72 (t, 1H, $J=10.0$ Hz), 3.60–3.65 (m, 2H), 3.58 (dd, 1H, J=4.0, 11.0 Hz), 3.54 (dd, 1H, J=2.0, 11.0 Hz), 3.39–3.42 (m, 1H), 3.36 (s, 3H), 3.27 (dd, 1H, $J=8.0$, 12.0 Hz); ¹³C NMR (125 MHz, CDCl₃) δ 159.23, 138.24, 137.41, 137.24, 137.17, 135.73, 129.49, 128.98, 128.59, 128.50, 128.32, 128.20, 128.14, 127.94, 127.81, 127.43, 126.14, 101.66, 100.96, 97.86, 80.74, 80.58, 79.69, 77.16, 74.76, 74.19, 73.46, 72.94, 72.61, 68.97, 68.19, 62.28, 60.41, 55.28, 47.84; MS (FAB) 829 [M]⁺; Anal. Calcd for C₄₉H₅₁NO₁₁: C, 70.91; H, 6.19; N, 1.69. Found: C, 70.80; H, 6.25; N, 1.64.

4.3.3. Coupling of 1e with **6a** to give methyl (N-acetyl-2-amino-2,3-N,O-carbonyl-4,6-dibenzyl-2-deoxy-p-glucopyranosyl)- $(1\rightarrow3)$ -2-O-benzyl-4,6-O-benzylidene- α -D-glucopyranoside (7e)

Donor $1e$ was pre-activated at -60 °C. The crude product was purified by column chromatography (petroleum ether/ethyl acetate, 4:1) to give 7e as an inseparable anomeric mixture, and anomeric ratio was determined by integration of the ¹H NMR spectrum of the reaction mixture, yield=84% (α/β =1:1). R_f=0.30 (petroleum ether/ethyl acetate, 1.5:1); 1 H NMR (500 MHz, CDCl3) δ 7.40–7.46 (m, 4H), 7.25–7.36 (m, 20H), 7.14 (t, 1H, J=7.0 Hz), 7.00 (t, 2H, $J = 7.5$ Hz), 6.65 (d, 2H, $J = 7.5$ Hz), 5.52 (s, 0.8H), 5.40 (s, 1H), 5.36 $(d, 1H, J=3.0 Hz, H-1/\alpha)$, 5.08 (d, 0.7H, J=7.5 Hz, H-1 β), 4.84 (d, 1H, $J=12.0$ Hz), 4.81 (d, 0.8H, $J=12.0$ Hz), 4.72 (d, 1.7H, $J=4.0$ Hz, H-1), 4.49–4.67 (m, 8H), 4.23–4.34 (m, 4H), 4.13–4.15 (m, 2H), 4.03 (d, 2H, $J=4.0$ Hz), 4.00 (dd, 0.8H, $J=10.0$, 12.0 Hz), 3.80–3.88 (m, 1.8H), 3.53–3.77 (m, 8H), 3.44–3.48 (m, 1H), 3.41 (s, 3H), 3.36 (s, 2.2H), 3.25 (dd, 0.7H, J=7.5, 11.5 Hz, H-2 β), 3.01 (dd, 1H, J=3.0, 12.0 Hz, H- 2α), 1.88 (s, 3H), 1.80 (s, 2H); HRMS (ESI) calcd for C₄₄H₄₇NO₁₂Na $[M+Na]^+$: 804.2990, found: 804.2994.

4.3.4. Coupling of 1g with 6a to give methyl (N-benzyl-2-amino-2,3-N,O-carbonyl-4,6-diacetyl-2-deoxy- α - and β -Dglucopyranosyl)-(1 \rightarrow 3)-2-O-benzyl-4,6-O-benzylidene- α -Dglucopyranosides ($7g\alpha$) and ($7g\beta$)

Donor $1g$ was pre-activated at -50 °C. The crude product was purified by column chromatography (petroleum ether/ethyl acetate, 7:1) to give **7g**, yield=94% (α/β =3:1). $R_f(\alpha$ -isomer)=0.45, $R_f(\beta - \alpha)$ isomer)=0.40 (petroleum ether/ethyl acetate, 1:1); compound $7g\alpha$: ¹H NMR (500 MHz, CDCl₃) δ 7.42–7.44 (m, 2H), 7.31–7.39 (m, 8H), 7.13–7.16 (m, 1H), 6.97–7.00 (m, 2H), 6.61–6.63 (m, 2H), 5.44 (s, 1H), 5.38 (d, 1H, $J=3.0$ Hz, H-1 \prime), 5.06 (t, 1H, $J=10.0$ Hz), 4.78 (d, 1H, $J=4.0$ Hz, H-1), 4.67 (d, 1H, $J=15.0$ Hz), 4.65 (d, 1H, $J=11.0$ Hz), 4.57 (d, 1H, $J=11.0$ Hz), 4.50 (dd, 1H, $J=10.0$, 12.0 Hz), 4.29 (t, 1H, J=9.5 Hz), 4.26 (dd, 1H, J=5.0, 10.5 Hz), 4.15-4.18 (m, 1H), 3.92 (dd, 1H, $J=2.0$, 12.5 Hz), 3.86–3.90 (m, 1H), 3.83 (dd, 1H, $J=4.0$, 12.5 Hz), 3.71 (t, 1H, J = 10.5 Hz), 3.64 (d, 1H, J = 15.0 Hz), 3.60 (t, 1H, J = 9.5 Hz), 3.58 (dd, 1H, J=3.5, 9.5 Hz), 3.44 (s, 3H), 3.09 (dd, 1H, J=3.0, 12.0 Hz), 2.03 (s, 3H), 2.01 (s, 3H); 13C NMR (125 MHz, CDCl3) d 170.58, 169.09, 158.09, 137.13, 136.85, 134.34, 129.80, 128.64, 128.58, 128.46, 128.36, 128.30, 127.78, 126.66, 102.75, 98.04, 94.52, 82.19, 77.58, 73.73, 73.54, 72.55, 70.15, 69.01, 68.00, 61.93, 61.06, 59.23, 55.39, 46.86, 20.62 (2C); HRMS (ESI) calcd for C₃₉H₄₃NO₁₃Na $[M+Na]^+$: 756.2627, found: 756.2629. Compound 7g β : ¹H NMR $(500 \text{ MHz}, \text{CDCl}_3)$ δ 7.45–7.47 (m, 2H), 7.41–7.43 (m, 2H), 7.28–7.37 (m, 11H), 5.54 (s, 1H), 5.13 (dd, 1H, J=8.5, 10.5 Hz), 5.10 (d, 1H, J=7.5 Hz, H-1^t), 4.74 (d, 1H, J=3.5 Hz, H-1), 4.56 (d, 1H, J=14.5 Hz), 4.53 (s, 2H), 4.32 (t, 1H, J=9.5 Hz), 4.30 (d, 1H, J=14.5 Hz), 4.27 (dd, 1H, $J=4.5$, 10.0 Hz), 4.08 (dd, 1H, $J=5.0$, 12.5 Hz), 4.01 (dd, 1H, $J=2.5$, 12.5 Hz), 3.96 (dd, 1H, J=10.0, 12.0 Hz), 3.84 (dt, 1H, J=4.5, 10.0 Hz), 3.76 (t, 1H, $=$ 10.0 Hz), 3.68 (dd, 1H, $=$ 3.5, 9.5 Hz), 3.58 (t, 1H, $J=9.5$ Hz), 3.52–3.55 (m, 1H), 3.33–3.38 (m, 1H), 3.36 (s, 3H), 2.05 (s, 3H), 1.93 (s, 3H); ¹³C NMR (125 MHz, CDCl₃) δ 170.58, 169.15, 158.39, 137.14, 135.22, 129.52, 129.10, 128.64, 128.42, 128.18, 128.12, 128.02, 126.04, 101.63, 101.34, 97.79, 80.42, 79.37, 75.86, 74.26, 72.60, 68.90, 67.68, 62.35, 62.10, 60.37, 60.03, 55.30, 47.84, 21.03, 20.64; HRMS (ESI) calcd for $C_{39}H_{43}NO_{13} + NH_4$ [M+NH₄]⁺: 751.3073, found: 751.3074.

4.4. Glycosylations of 1c with 6a–6m by means of BSM/Tf_2O pre-activation

General procedure: Triflic anhydride (11.8 µL, 0.070 mmol, 1.3 equiv) was added to a stirred solution of 1c (37.6 mg, 0.065 mmol, 1.2 equiv), BSM (15.9 mg, 0.075 mmol, 1.4 equiv) and activated 4 Å molecular sieves (500 mg) in CH₂Cl₂ (6.0 mL) at -73 °C under nitrogen atmosphere. The reaction mixture was stirred for 3–5 min, after disappearance of 1c detected by TLC, a solution of the acceptor alcohol **6a** (25.0 mg, 0.054 mmol, 1.0 equiv) or **6b-6m** in CH_2Cl_2 (0.5 mL) was added dropwise to the pre-activated system. The mixture was stirred and slowly warmed to room temperature, and then quenched by Et_3N (0.1 mL). The precipitate was filtered off and the filtrate was concentrated. The crude product was purified by column chromatography on silica gel to give 7c and 8b–8m.

4.4.1. Coupling of 1c with 6b to give methyl (N-benzyl-2-amino-2,3-N,O-carbonyl-4,6-dibenzyl-2-deoxy- β -D-glucopyranosyl)-(1 \rightarrow 2)-3-O-benzyl-4,6-O-benzylidene- α -D-glucopyranoside (8b)

The crude product was purified by column chromatography (petroleum ether/ethyl acetate, 7:1) to give **8b**, yield=83%. R_f =0.50 (petroleum ether/ethyl acetate, 2:1); $\mathrm{^{1}H}$ NMR (500 MHz, CDCl₃) d 7.58–7.60 (m, 2H), 7.48–7.50 (m, 2H), 7.24–7.40 (m, 19H), 7.18–7.20 $(m, 2H)$, 5.59 (s, 1H), 5.04 (d, 1H, $I=8.0$ Hz, H-1 \prime), 5.00 (d, 1H, $J=11.0$ Hz), 4.85 (d, 1H, J=3.5 Hz, H-1), 4.80 (d, 1H, J=11.5 Hz), 4.63 (d, 1H, J = 14.5 Hz), 4.56 (d, 1H, J = 11.0 Hz), 4.52 (d, 1H, J = 12.5 Hz), 4.45 (d, 1H, J=12.5 Hz), 4.44 (d, 1H, J=11.5 Hz), 4.34 (dd, 1H, J=5.0, 10.5 Hz), 4.20 (t, 1H, J=9.5 Hz), 4.18 (d, 1H, J=14.5 Hz), 3.88–3.97 (m, 3H), 3.78 (t, 1H, J=10.0 Hz), 3.71 (dd, 1H, J=3.0, 9.0 Hz), 3.60–3.68 $(m, 3H)$, 3.49 (s, 3H), 3.45–3.48 $(m, 1H)$, 3.14 (dd, 1H, J=7.5, 12.0 Hz); ¹³C NMR (125 MHz, CDCl₃) δ 158.91, 137.93, 137.82, 137.25 (2C), 135.24, 130.47, 129.02, 128.50, 128.40, 128.37, 128.28, 128.04, 128.00, 127.94, 127.91, 127.74, 127.71, 125.95, 102.06, 101.40, 100.09, 83.19, 80.62, 78.43, 76.69, 76.17, 75.18, 73.93, 73.36, 73.00, 69.16, 68.31, 62.20, 59.63, 55.45, 47.68; MS (FAB) 829 [M]⁺, 830 [M+H]⁺; Anal. Calcd for $C_{49}H_{51}NO_{11}$: C, 70.91; H, 6.19; N, 1.69. Found: C, 71.11; H, 6.32; N, 1.50.

4.4.2. Coupling of 1c with 6c to give methyl (N-benzyl-2-amino-2,3-N,O-carbonyl-4,6-dibenzyl-2-deoxy- β -D-glucopyranosyl)-(1 \rightarrow 3)-2-O-benzyl-4,6-O-benzylidene- α -D-galactopyranoside (8c)

The crude product was purified by column chromatography (petroleum ether/ethyl acetate, 3:1) to give **8c**, yield=83%. R_f =0.20 (petroleum ether/ethyl acetate, 2:1); 1 H NMR (500 MHz, CDCl₃) δ 7.56–7.58 (m, 2H), 7.42–7.43 (m, 2H), 7.24–7.38 (m, 16H), 7.18–7.20 $(m, 2H)$, 7.10 (t, 1H, J=7.5 Hz), 6.85 (t, 2H, J=7.5 Hz), 5.55 (s, 1H), 5.08 $(d, 1H, J=8.0 Hz, H-1)$, 4.89 $(d, 1H, J=3.5 Hz, H-1)$, 4.80 $(d, 1H, J=4.80)$ $J=11.0$ Hz), 4.59 (d, 2H, $J=12.5$ Hz), 4.55 (d, 1H, $J=12.0$ Hz), 4.53 (d, 1H, $J=11.5$ Hz), 4.44 (d, 1H, $J=12.0$ Hz), 4.43 (d, 1H, $J=11.5$ Hz), 4.34 $(d, 1H, J=3.5 Hz)$, 4.28 (dd, 1H, J=3.5, 10.0 Hz), 4.21–4.26 (m, 2H), 4.19 (dd, 1H, J=3.5, 10.5 Hz), 4.00 (dd, 1H, J=1.5, 12.5 Hz), 3.98 (dd, 1H, J = 10.0, 12.0 Hz), 3.72 (dd, 1H, J = 9.0, 10.0 Hz), 3.64–3.67 (m, 3H), 3.53–3.56 (m, 1H), 3.38 (s, 3H), 3.18 (dd, 1H, J=8.0, 12.0 Hz); ¹³C NMR (125 MHz, CDCl₃) δ 158.98, 137.96, 137.89, 137.50, 137.28, 135.10, 130.20, 129.11, 128.56, 128.36, 128.32, 128.23, 128.04, 128.02, 127.89, 127.68, 126.46, 102.75, 101.34, 98.16, 80.71, 76.75, 76.60, 76.59, 74.04, 73.76, 73.35, 72.97, 72.76, 69.23, 68.48, 62.54, 59.70, 55.47, 47.69; MS (FAB) 852 [M+Na]⁺; Anal. Calcd for C₄₉H₅₁NO₁₁: C, 70.91; H, 6.19; N, 1.69. Found: C, 71.05; H, 6.66; N, 1.49.

4.4.3. Coupling of 1c with 6d to give methyl (N-benzyl-2-amino-2,3-N,O-carbonyl-4,6-dibenzyl-2-deoxy- β -D-glucopyranosyl)-(1 \rightarrow 2)-3-O-benzyl-4,6-O-benzylidene- α -D-galactopyranoside (8d)

The crude product was purified by column chromatography (petroleum ether/ethyl acetate, 5:1) to give 8d, yield=83%. R_f =0.3 (petroleum ether/ethyl acetate, 2:1); $\mathrm{^{1}H}$ NMR (500 MHz, CDCl₃) δ 7.53–7.59 (m, 4H), 7.23–7.36 (m, 18H), 7.17–7.19 (m, 2H), 5.46 (s, 1H), 5.02 (d, 1H, J=3.5 Hz, H-1), 4.97 (d, 1H, J=7.5 Hz, H-1^t), 4.79 (d, 1H, J = 11.5 Hz), 4.59 (d, 1H, J = 15.0 Hz), 4.54 (d, 1H, J = 11.5 Hz), 4.52 $(d, 1H, J=12.0 Hz)$, 4.43 $(d, 1H, J=10.0 Hz)$, 4.42 $(d, 1H, J=12.5 Hz)$, 4.28–4.31 (m, 3H), 4.27 (d, 1H, $J=14.5$ Hz), 4.12 (dd, 1H, $J=3.5$, 10.0 Hz), 4.05 (dd, 1H, J=1.5, 12.5 Hz), 3.97 (dd, 1H, J=10.0, 12.0 Hz), 3.76 (t, 1H, J=9.5 Hz), 3.69 (s, 1H), 3.66 (dd, 1H, J=4.5, 11.5 Hz), 3.61 $(dd, 1H, J=1.5, 10.5 Hz$), 3.47–3.50 (m, 1H), 3.46 (s, 3H), 3.21 (dd, 1H, $J=8.0$, 11.5 Hz); ¹³C NMR (125 MHz, CDCl₃) δ 158.86, 137.86, 137.64, 137.60, 137.36, 135.36, 130.40, 128.92, 128.40, 128.33, 128.29, 128.13, 128.02, 127.93, 127.90, 128.81, 127.77, 127.71, 127.63, 126.20, 102.44, 101.10, 100.19, 80.73, 76.51, 75.94, 74.38, 73.82, 73.42, 73.25, 72.90, 70.89, 69.49, 68.17, 62.25, 59.54, 55.71, 47.31; MS (FAB) 852 [M+Na]⁺, 868 [M+K]⁺; Anal. Calcd for C₄₉H₅₁NO₁₁: C, 70.91; H, 6.19; N, 1.69. Found: C, 70.73; H, 6.18; N, 1.53.

4.4.4. Coupling of 1c with 6e to give methyl (N-benzyl-2-amino-2,3-N,O-carbonyl-4,6-dibenzyl-2-deoxy- β -D-glucopyranosyl)-(1 \rightarrow 3)-2-O-benzyl-4,6-O-benzylidene- β -D-glucopyranoside (8e)

The crude product was purified by column chromatography (petroleum ether/ethyl acetate, 8:1) to give **8e**, yield=87%. R_f =0.30 (petroleum ether/ethyl acetate, 3:1); 1 H NMR (500 MHz, CDCl₃) δ 7.40–7.43 (m, 4H), 7.16–7.32 (m, 21H), 5.49 (s, 1H), 5.04 (d, 1H, $J=7.5$ Hz, H-1 ℓ), 4.85 (d, 1H, J=11.5 Hz), 4.78 (d, 1H, J=11.0 Hz), 4.49 (d, 1H, J = 14.5 Hz), 4.45 (d, 1H, J = 11.5 Hz), 4.40 (d, 1H, J = 7.5 Hz, H-1), 4.39 (d, 1H, J=11.0 Hz), 4.31–4.37 (m, 4H), 4.09 (t, 1H, J=9.5 Hz), 3.88 (dd, 1H, $J=9.5$, 12.0 Hz), 3.77 (t, 1H, $J=10.5$ Hz), 3.72 (dd, 1H, $J=8.5, 9.5$ Hz), 3.66 (t, 1H, $J=9.5$ Hz), 3.57–3.59 (m, 1H), 3.56 (s, 3H), 3.54 (dd, 1H, J=4.5, 11.0 Hz), 3.49 (dd, 1H, J=7.5, 8.0 Hz), 3.35–3.43 (m, 2H), 3.23 (dd, 1H, J=7.5, 11.5 Hz); ¹³C NMR (125 MHz, CDCl₃) d 159.20, 138.17, 137.87, 137.36, 137.12, 135.81, 129.32, 129.01, 128.50, 128.36, 128.26, 128.17, 127.97, 127.89, 127.85, 127.78, 127.51, 126.14, 105.08, 101.49, 100.34, 82.61, 80.61, 78.98, 77.17, 74.44, 74.21, 73.43, 72.90, 68.74, 68.47, 66.15, 60.51, 57.28, 47.90; MS (FAB) 853 $[M+Na+H]^+$; Anal. Calcd for C₄₉H₅₁NO₁₁: C, 70.91; H, 6.19; N, 1.69. Found: C, 71.11; H, 6.29; N, 1.42.

4.4.5. Coupling of 1c with 6f to give methyl (N-benzyl-2-amino-2,3-N,O-carbonyl-4,6-dibenzyl-2-deoxy-p-glucopyranosyl)-(1 \rightarrow 2)-3-O-benzyl-4,6-O-benzylidene- β -D-glucopyranoside (8f)

The crude product was purified by column chromatography (petroleum ether/ethyl acetate, 4:1) to give 8f as an inseparable anomeric mixture, and anomeric ratio was determined by integration of the ¹H NMR spectrum of the reaction mixture, yield=85% (α/β =1:2). R=0.30 (petroleum ether/ethyl acetate, 2:1); ¹H NMR (500 MHz, CDCl₃) δ 7.46–7.49 (m, 3H), 7.11–7.40 (m, 33H), 7.03–7.06 (m, 2H), 5.59 (s, 0.5H), 5.58 (s, 1H), 5.52 (d, 0.6H, J=3.0 Hz, H-1a \prime), 5.06 (d, 1H, J=7.5 Hz, H-1b \prime), 4.96 (d, 1H, J=11.0 Hz), 4.93 (d, 0.6H, $J=10.5$ Hz), 4.81–4.87 (m, 2H), 4.59 (d, 1H, $J=10.5$ Hz), 4.57 (d, 1H, J=12.0 Hz), 4.44-4.51 (m, 5.5H), 4.37-4.41 (m, 2H), 4.33 (d, 0.7H, J=7.0 Hz, H-1a), 4.27 (d, 1H, J=14.5 Hz), 4.18 (d, 0.6H, J¼12.0 Hz), 3.67–4.02 (m, 12H), 3.53 (s, 1.7H), 3.46 (s, 3H), 3.43–3.49 (m, 2H), 3.20–3.26 (m, 2H), 3.13 (dd, 0.7H); MS (FAB) 853 [M+Na+H]⁺; Anal. Calcd for C₄₉H₅₁NO₁₁: C, 70.91; H, 6.19; N, 1.69. Found: C, 71.65; H, 6.66; N, 1.44.

4.4.6. Coupling of 1c with 6g to give methyl (N-benzyl-2-amino-2,3-N,O-carbonyl-4,6-dibenzyl-2-deoxy- α - and β -D-

glucopyranosyl)-(1 \rightarrow 3)-2-azide-4,6-O-benzylidene-2-deoxy- α -Dmannopyranosides ($8g\alpha$) and ($8g\beta$)

The crude product was purified by column chromatography (petroleum ether/ethyl acetate, 7:1) to give 8g, yield=83% (α / β =1:1.5). $R_f(\alpha$ -isomer)=0.45, $R_f(\beta$ -isomer)=0.40 (petroleum ether/ ethyl acetate, 2:1); compound $\text{g}g\alpha$: ¹H NMR (500 MHz, CDCl₃) δ 7.44–7.46 (m, 2H), 7.25–7.34 (m, 11H), 7.15–7.21 (m, 3H), 7.06 (t, 2H, $J=8.0$ Hz), 6.72 (d, 2H, $J=7.5$ Hz), 5.49 (s, 1H), 5.29 (d, 1H, $J=2.5$ Hz, H-1 \prime), 4.85 (d, 1H, $J=11.5$ Hz), 4.59–4.65 (m, 3H), 4.53 (d, 1H, J = 12.0 Hz), 4.45 (d, 1H, J = 11.0 Hz), 4.43 (d, 1H, J = 12.5 Hz), 4.42 (dd, 1H, $J=3.5$, 10.0 Hz), 4.23-4.25 (m, 1H), 4.05 (t, 1H, $J=9.0$ Hz), 3.94 (dd, 1H, $J=1.5$, 3.5 Hz), 3.74-3.84 (m, 5H), 3.68 (d, 2H, J=2.5 Hz), 3.37 (s, 3H), 3.13 (dd, 1H, J=3.0, 12.5 Hz); ¹³C NMR (125 MHz, CDCl3) d 158.61, 137.72, 137.40, 136.95, 134.80, 129.73, 128.63, 128.55, 128.36, 128.14, 128.02, 127.81, 127.74, 126.56, 102.93, 99.89, 96.16, 78.56, 76.90, 74.56, 74.28, 73.53, 73.48, 72.81, 68.71,

67.97, 63.80, 63.70, 59.06, 55.13, 46.62; MS (FAB) 764 $[M]^{+}$, 765 $[M+H]^+$; Anal. Calcd for C₄₂H₄₄N₄O₁₀: C, 65.96; H, 5.80; N, 7.33. Found: C, 65.69; H, 6.04; N, 7.09. Compound $8gf:$ ¹H NMR (500 MHz, CDCl3) d 7.21–7.38 (m, 20H), 5.44 (s, 1H), 4.84 (d, 1H, J=8.0 Hz, H-1[']), 4.81 (d, 1H, J=11.5 Hz), 4.69 (d, 1H, J=1.5 Hz, H-1), 4.48–4.56 (m, 3H), 4.43 (s, 2H), 4.16–4.18 (m, 1H), 4.13 (dd, 1H, $J=4.0, 9.5$ Hz), $4.03-4.09$ (m, 2H), 3.70–3.79 (m, 4H), 3.50–3.55 (m, 2H), 3.37–3.41 (m, 1H), 3.35 (s, 3H), 3.32 (dd, 1H, J=4.0, 12.0 Hz); ^{13}C NMR (125 MHz, CDCl₃) δ 159.05, 138.07, 137.30, 137.16, 136.23, 129.22, 128.63, 128.36, 128.29, 128.17, 128.00, 127.90, 127.76, 127.64, 126.17, 102.04, 99.61, 98.70, 80.36, 77.54, 77.26, 74.01, 73.26, 73.17, 72.85, 68.62, 68.39, 63.54, 61.44, 60.96, 55.09, 47.90; MS (FAB) 764 [M]⁺, 765 [M+H]⁺; Anal. Calcd for C₄₂H₄₄N₄O₁₀: C, 65.96; H, 5.80; N, 7.33. Found: C, 65.69; H, 6.08; N, 7.10.

4.4.7. Coupling of **1c** with **6h** to give methyl (N-benzyl-2-amino-2,3-N,O-carbonyl-4,6-dibenzyl-2-deoxy- α - and β -Dglucopyranosyl)- $(1\rightarrow 2)$ -3-azide-4,6-O-benzylidene-3-deoxy- α -Dglucopyranosides ($8h\alpha$) and ($8h\beta$)

The crude product was purified by column chromatography (petroleum ether/ethyl acetate, 4:1) to give **8h**, yield=83% (α / β =1:2). R_f (β -isomer)=0.40, R_f (α -isomer)=0.25 (petroleum ether/ ethyl acetate, 2:1); compound $8h\alpha$: ¹H NMR (500 MHz, CDCl₃) d 7.48–7.50 (m, 2H), 7.21–7.38 (m, 18H), 5.56 (s, 1H), 5.09 (d, 1H, J=3.0 Hz, H-1^t), 4.93 (d, 1H, J=15.0 Hz), 4.89 (d, 1H, J=3.5 Hz, H-1), 4.87 (d, 1H, J=11.0 Hz), 4.68–4.72 (m, 1H), 4.56 (d, 1H, J=12.5 Hz), 4.48 (d, 1H, J=11.5 Hz), 4.42 (d, 1H, J=12.0 Hz), 4.32 (dd, 1H, J=5.0, 10.5 Hz), 4.01 (t, 1H, J=10.0 Hz), 3.98 (d, 1H, J=15.0 Hz), 3.91-3.94 $(m, 2H)$, 3.85 (dt, 1H, $J=5.0$, 10.0 Hz), 3.75 (dd, 1H, $J=2.5$, 10.5 Hz), 3.73 (t, 1H, $J=10.5$ Hz), 3.59–3.63 (m, 2H), 3.50 (s, 3H), 3.46 (t, 1H, $J=10.0$ Hz), 3.30 (dd, 1H, J=3.0, 12.0 Hz); ¹³C NMR (125 MHz, CDCl₃) d 158.48, 137.64, 137.44, 136.65, 134.95, 129.16, 128.41, 128.37, 128.32, 128.07, 127.91, 127.82, 125.98, 101.59, 96.38, 91.84, 80.01, 77.39, 74.41, 73.80, 73.45, 73.38, 72.97, 68.81, 67.43, 62.48, 61.04, 59.05, 55.11, 47.12; HRMS (ESI) calcd for $C_{42}H_{48}N_4O_{10}$ [M+NH₄]⁺: 782.3396, found: 728.3381. Compound $8h\beta$: ¹H NMR (500 MHz, CDCl₃) δ 7.48–7.54 (m, 4H), 7.21–7.40 (m, 16H), 5.60 (s, 1H), 4.92 (d, 1H, J = 7.5 Hz, H-1 \prime), 4.84 (d, 1H, J = 12.5 Hz), 4.83 (d, 1H, J = 4.0 Hz, H-1), 4.79 (d, 1H, $J=15.0$ Hz), 4.52 (d, 1H, $J=12.0$ Hz), 4.48 (d, 1H, $J=12.0$ Hz), 4.46 (d, 1H, J=12.5 Hz), 4.34 (d, 1H, J=14.5 Hz), 4.33 (dd, 1H, $J=5.0$, 10.0 Hz), 4.09-4.14 (m, 2H), 3.92 (dt, 1H, $J=5.0$, 10.0 Hz), 3.56–3.77 (m, 7H), 3.43 (s, 3H), 3.24 (dd, 1H, J=7.5, 11.5 Hz); ¹³C NMR (125 MHz, CDCl₃) δ 158.85, 137.74, 137.19, 136.71, 135.32, 129.85, 129.07, 128.47, 128.42, 128.28, 128.06, 127.96, 127.89, 127.81, 127.71, 125.89, 102.24, 101.46, 99.60, 80.90, 80.42, 76.91, 76.60, 73.80, 73.38, 72.98, 69.06, 68.53, 62.37, 61.44, 59.89, 55.64, 47.67; HRMS (ESI) calcd for $C_{42}H_{48}N_5O_{10}$ [M+NH₄]⁺: 782.3396, found: 782.3384.

4.4.8. Coupling of 1c with 6i to give N-benzyl-2-amino-2,3-N,Ocarbonyl-4,6-dibenzyl-2-deoxy-p-glucopyranosyl- $(1\rightarrow6)$ -1,2:3,4di-O-isopropylidene- α -D-galactopyranoside (8i)

The crude product was purified by column chromatography (petroleum ether/ethyl acetate, 5:1) to give $8i$ as an inseparable anomeric mixture, and anomeric ratio was determined by integration of the 1 H NMR spectrum of the reaction mixture, yield=96% (α/β =1:11). R_f=0.35 (petroleum ether/ethyl acetate, 2:1); major isomer $8i\beta$: ¹H NMR (500 MHz, CDCl₃) δ 7.54–7.55 (m, 2H), 7.25–7.53 (m, 11H), 7.17–7.19 (m, 2H), 5.58 (d, 1H, $J=5.0$ Hz, H-1), 4.81 (d, 1H, J = 11.0 Hz), 4.71 (d, 1H, J = 14.5 Hz), 4.70 (d, 1H, J = 8.0 Hz, H-1 \prime), 4.61 (d, 1H, J=5.0, 8.0 Hz), 4.57 (d, 1H, J=12.5 Hz), 4.46 (d, 1H, $J=11.0$ Hz), 4.43 (d, 1H, $J=11.0$ Hz), 4.41 (d, 1H, $J=12.5$ Hz), 4.22 (dd, 1H, $J=2.0$, 5.0 Hz), 4.19 (dd, 1H, $J=2.0$, 8.0 Hz), 4.15 (dd, 1H, $J=2.0$, 11.0 Hz), 4.08–4.13 (m, 2H), 4.07 (dd, 1H, J=9.5, 12.0 Hz), 3.79 (dd, 1H, $J=8.5$, 9.5 Hz), 3.72–3.76 (m, 1H), 3.69 (dd, 1H, $J=3.0$, 11.5 Hz), 3.53–3.56 (m, 1H), 3.22 (dd, 1H, J=8.0, 12.0 Hz), 1.48 (s, 3H), 1.47 (s, 3H), 1.32 (s, 3H), 1.29 (s, 3H); MS (FAB) 718 $[M+H]^+$; Anal. Calcd for C40H47NO11: C, 66.93; H, 6.60; N, 1.95. Found: C, 67.02; H, 6.72; N, 1.91.

4.4.9. Coupling of 1c with 6j to give methyl (N-benzyl-2-amino-2,3- $N, O-carbonyl-4, 6-dibenzyl-2-deoxy-p-glucopy ranosyl)-(1\rightarrow6)-$ 2,3,4-tri-O-benzyl- α -D-glucopyranoside (8j)

The crude product was purified by column chromatography (petroleum ether/ethyl acetate, 4:1) to give $8j$ as an inseparable anomeric mixture, and anomeric ratio was determined by integration of the $^1\mathrm{H}$ NMR spectrum of the reaction mixture, yield= 85% ($\alpha/\beta=1:6$). R_f=0.25 (petroleum ether/ethyl acetate, 2:1); major isomer **8j**β: ¹H NMR (500 MHz, CDCl₃) δ 7.42–7.44 (m, 2H), 7.17– 7.35 (m, 28H), 4.98 (d, 1H, $J=11.0$ Hz), 4.85 (d, 1H, $J=11.5$ Hz), 4.81 (d, 1H, J = 11.5 Hz), 4.80 (d, 1H, J = 10.5 Hz), 4.78 (d, 1H, J = 11.0 Hz), 4.62 (d, 1H, J=12.0 Hz), 4.58 (d, 1H, J=15.0 Hz), 4.55 (d, 1H, J=3.5 Hz, H-1), 4.54 (d, 1H, J=10.5 Hz), 4.52 (d, 1H, J=10.0 Hz), 4.48 (d, 1H, J=15.0 Hz), 4.45 (d, 1H, J=11.0 Hz), 4.38 (d, 1H, J=7.5 Hz, H-1⁻), 4.31 (d, 1H, J = 14.5 Hz), 4.01 (t, 1H, J = 12.0 Hz), 3.99–4.04 (m, 2H), 3.82– 3.85 (m, 1H), 3.73 (dd, 1H, J=8.5, 9.5 Hz), 3.62–3.66 (m, 2H), 3.51 $(dd, 1H, J=5.5, 11.0 Hz$), 3.46 (dd, 1H, J=3.5, 10.0 Hz), 3.47–3.50 (m, 1H), 3.34 (dd, 1H, J=9.0, 10.0 Hz), 3.33 (s, 3H), 3.22 (dd, 1H, J=7.5, 12.0 Hz); MS (ESI) 939 $[M+NH_4]^+$, 944 $[M+Na]^+$, 960 $[M+K]^+$; Anal. Calcd for $C_{56}H_{59}NO_{11}$: C, 72.94; H, 6.45; N, 1.52. Found: C, 72.67; H, 6.44; N, 1.55.

4.4.10. Coupling of 1c with 6k to give N-benzyl-2-amino-2,3-N,Ocarbonyl-4,6-dibenzyl-1-O-octyl-2-deoxy- α - and β - D glucopyranosides ($8k\alpha$) and ($8k\beta$)

The crude product was purified by column chromatography (petroleum ether/ethyl acetate, 11:1 to 9:1) to give $8k$, yield=95% $(\alpha/\beta=1:3.5)$. R_f (β -isomer)=0.7, R_f (α -isomer)=0.5 (petroleum ether/ ethyl acetate, 2:1); compound $8k\alpha$: ¹H NMR (500 MHz, CDCl₃) δ 7.20–7.35 (m, 14H), 4.86 (d, 1H, J=11.0 Hz), 4.68 (d, 1H, J=3.0 Hz, H-1), 4.64 (dd, 1H, J=10.0, 12.0 Hz), 4.56 (d, 1H, J=12.5 Hz), 4.47 (d, 1H, $J=11.0$ Hz), 4.42 (d, 1H, $J=12.0$ Hz), 4.41 (s, 2H), 3.90 (t, 1H, $J=9.0$ Hz), 3.72 (dd, 1H, $J=3.5$, 10.5 Hz), 3.65–3.67 (m, 1H), 3.59 (dd, 1H, $J=1.5$, 10.5 Hz), 3.46–3.51 (m, 1H), 3.31 (dd, 1H, $J=3.0$, 12.0 Hz), 3.00–3.05 (m, 1H), 1.46 (s, 2H), 1.26–1.32 (m, 10H), 0.88 (t, 3H, J=6.5 Hz); ¹³C NMR (125 MHz, CDCl₃) δ 159.14, 137.78, 137.45, 135.31, 128.75, 128.63, 128.36, 128.13, 128.04, 127.84, 127.75, 95.17, 77.38, 74.88, 73.46, 73.00, 72.42, 68.65, 67.72, 61.24, 48.64, 31.80, 29.31, 29.17, 26.06, 22.62, 14.07; MS (ESI) 588 [M+H]⁺, 605 $[M+NH_4]^+$, 610 $[M+Na]^+$; Anal. Calcd for $C_{36}H_{45}NO_6$: C, 73.57; H, 7.72; N, 2.38. Found: C, 73.41; H, 7.47; N, 2.36. Compound **8k**β: ¹H NMR (500 MHz, CDCl3) d 7.39–7.41 (m, 2H), 7.25–7.34 (m, 11H), 7.21–7.23 (m, 2H), 4.84 (d, 1H, J=11.0 Hz), 4.57 (d, 1H, J=12.0 Hz), 4.56 (d, 1H, J=7.5 Hz, H-1), 4.51 (d, 1H, J=15.0 Hz), 4.50 (d, 1H, J=12.5 Hz), 4.49 (d, 1H, J=15.0 Hz), 4.46 (d, 1H, J=12.0 Hz), 4.11 (dd, 1H, J=9.5, 12.0 Hz), 3.81-3.86 (m, 1H), 3.79 (dd, 1H, J=8.0, 9.5 Hz), 3.71 (dd, 1H, $I=2.0$, 11.0 Hz), 3.67 (dd, 1H, $I=5.0$, 11.0 Hz), 3.55–3.58 $(m, 1H)$, 3.28–3.33 $(m, 1H)$, 3.26 $(dd, 1H, J=7.0, 12.0 Hz)$, 1.55–1.57 (m, 2H), 1.26–1.30 (m, 10H), 0.88 (t, 3H, J=7.0 Hz); ¹³C NMR (125 MHz, CDCl3) d 159.13, 137.98, 137.33, 136.15, 128.80, 128.47, 128.39, 128.01, 127.91, 127.74, 127.68, 101.59, 80.85, 76.75, 74.17, 73.45, 73.00, 69.78, 68.52, 60.72, 47.84, 31.78, 29.47, 29.32, 29.21, 26.04, 22.63, 14.08; MS (FAB) 587 [M]⁺, 588 [M+H]⁺; Anal. Calcd for $C_{36}H_{45}NO_6$: C, 73.57; H, 7.72; N, 2.38. Found: C, 73.79; H, 8.00; N, 2.58.

4.4.11. Coupling of 1c with 6l to give N-benzyl-2-amino-2,3-N,Ocarbonyl-4,6-dibenzyl-1-O-(N-carbobenzyloxy-L-serine methyl ester)-2-deoxy-_D-glucopyranoside (8l)

The crude product was purified by column chromatography (petroleum ether/ethyl acetate, 3:1) to give 8l as an inseparable anomeric mixture, and anomeric ratio was determined by

integration of the 1 H NMR spectrum of the reaction mixture, yield=98% (α/β =1:3). Rf=0.25 (petroleum ether/ethyl acetate, 1:1); major isomer **8Ιβ**: ¹Η NMR (500 MHz, CDCl₃) δ 7.21–7.35 (m, 20Η), 5.53 (d, 1H, J=7.5 Hz), 5.10 (d, 1H, J=15.5 Hz), 5.09 (d, 1H, J=15.5 Hz), 4.83 (d, 1H, J=11.0 Hz), 4.51 (d, 1H, J=8.0 Hz, H-1), 4.40-4.49 (m, 5H), 4.03–4.09 (m, 2H), 3.73–3.82 (m, 2H), 3.70 (s, 3H), 3.63–3.65 $(m, 3H)$, 3.50–3.52 $(m, 1H)$, 3.21 $(dd, 1H, J=8.0, 12.0 Hz$; HRMS (ESI) calcd for $C_{40}H_{43}N_2O_{10}$ [M+H]⁺: 711.2912, found: 711.2976.

4.4.12. Coupling of 1c with 6m to give methyl (N-benzyl-2-amino-2,3-N,O-carbonyl-4,6-dibenzyl-2-deoxy-p-glucopyranosyl)- $(1\rightarrow4)$ -2,3,6-tri-O-benzyl- α -D-glucopyranoside (8m)

The crude product was purified by column chromatography (petroleum ether/ethyl acetate, 5:1) to give $8m$ as an inseparable anomeric mixture, and anomeric ratio was determined by integration of the ¹H NMR spectrum of the mixture, yield=82% (a/ β = 2.5:1). R_f =0.3 (petroleum ether/ethyl acetate, 2:1); ¹H NMR (500 MHz, CDCl₃) δ 7.03-7.35 (m, 56H), 5.54 (d, 1H, J=3.0 Hz, H- 1α), 5.08 (d, 1H, J=12.0 Hz), 5.02 (d, 0.7H, J=12.0 Hz), 3.87–4.05 (m, 6H), 3.73–3.84 (m, 4H), 3.67–3.70 (m, 1H), 3.41–3.63 (m, 8H), 3.38 $(s, 3H)$, 3.35 $(s, 3H)$, 3.27–3.30 (m, 1H), 3.17 (dd, 0.9H, J=3.0, 12.0 Hz, H-2 α), 3.01 (dd, 0.4H, J=7.5, 12.0 Hz, H-2 β); MS (FAB) 921 [M]⁺; Anal. Calcd for $C_{56}H_{59}NO_{11}$: C, 72.94; H, 6.45; N, 1.52. Found: C, 72.75; H, 6.29; N, 1.43.

4.5. Deprotection of disaccharide 8d

4.5.1. Methyl (N-benzyl-2-amino-4,6-O-dibenzyl-2-deoxy-β-Dglucopyranosyl)-(1 \rightarrow 2)-3-O-benzyl-4,6-O-benzylidene- α -Dgalactopyranoside (9)

Compound 8d (50.0 mg, 0.066 mmol, 1.0 equiv) was dissolved in DMSO (2.0 mL) and stirred, and then t-BuOK (34.0 mg, 0.30 mmol, 5.0 equiv) was added. The reaction mixture was stirred at room temperature for 20 min, diluted with ethyl acetate and washed with water. The aqueous layer was extracted with ethyl acetate $(3\times2$ mL). The combined organic layer was dried, filtered and concentrated under reduced pressure. The residue was purified by column chromatography (petroleum ether/ethyl acetate, 1:1) to give 9 (44 mg, 90%) as a white solid: 1 H NMR (500 MHz, CDCl₃) d 7.55–7.57 (m, 2H), 7.23–7.37 (m, 20H), 7.18–7.20 (m, 3H), 5.54 (s, 1H), 4.84 (d, 1H, J=11.0 Hz), 4.77 (d, 1H, J=3.5 Hz, H-1), 4.71 (d, 1H, J=8.0 Hz, H-1[']), 4.49–4.61 (m, 5H), 4.38 (d, 1H, J=3.5 Hz), 4.23 (dd, 1H, $J=3.5$, 10.5 Hz), 4.20 (dd, 1H, $J=1.0$, 12.5 Hz), 4.12 (dd, 1H, $J=3.0$, 10.0 Hz), 4.07 (d, 1H, J=13.0 Hz), 3.99 (d, 1H, J=13.0 Hz), 3.96 (dd, 1H, J=1.5, 12.5 Hz), 3.73 (dd, 1H, J=2.0, 11.0 Hz), 3.68 (dd, 1H, J=5.0, 11.0 Hz), 3.61 (s, 1H), 3.38–3.50 (m, 3H), 3.36 (s, 3H), 3.22 (br, 1H), 2.53 (dd, 1H, J=8.0, 9.0 Hz), 1.67 (br.s, 1H); ¹³C NMR (125 MHz, CDCl3) d 140.27, 138.37, 138.23, 138.03, 137.98, 128.72, 128.51, 128.33, 128.18, 128.02, 127.93, 127.76, 127.68, 127.62, 126.88, 126.18, 104.86, 100.60, 98.54, 78.10, 76.82, 75.73, 74.61, 74.24, 73.90, 73.37, 72.94, 69.26, 69.19, 62.70, 62.37, 55.43, 51.56, 42.65; HRMS (ESI) calcd for C₄₈H₅₄NO₁₀ [M+H]⁺: 804.3742, found: 804.3759.

4.5.2. Methyl (2-amino-2-deoxy- β -D-glucopyranosyl)-(1 \rightarrow 2)- α -Dgalactopyranoside (10)

Compound 9 (44.0 mg, 0.055 mmol) was dissolved in THF/water (2:1, 3.0 mL) containing HOAc (0.15 mL), and 10% Pd/C (15.0 mg) was added. The mixture was hydrogenated under an atmospheric pressure of hydrogen. After stirring for 24 h, the catalyst was removed by filtration and the filtrate was concentrated. The residue was subjected to a C-18 reverse phase column chromatography $(H₂O)$ to give **13** (18.0 mg, 95%) as a white solid after lyophilization: ¹H NMR (300 MHz, D₂O) δ 4.89 (d, 1H, J=3.5 Hz, H-1), 4.49 (d, 1H, J¼8.5 Hz, H-10), 3.72–3.82 (m, 5H), 3.56–3.62 (m, 3H), 3.25–3.31 (m, 3H), 3.23 (s, 3H), 2.65 (t, 1H, J=9.0 Hz); ¹³C NMR (125 MHz, D₂O) d 106.11, 101.55, 80.78, 78.61, 77.27, 73.18, 72.19, 71.98, 70.98, 63.77, 63.26, 59.04, 57.34; HRMS (ESI) calcd for $C_{13}H_{26}NO_{10}$ [M+H]⁺: 356.1551, found: 356.1554.

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Supplementary data

 $¹H$ and $¹³C$ NMR spectra of all new compounds. Supplementary</sup></sup> data associated with this article can be found in the online version, at [doi:10.1016/j.tet.2008.03.103](http://dx.doi.org/doi:10.1016/j.tet.2008.03.103).

References and notes

- 1. (a) Dwek, R. A. Chem. Rev. 1996, 96, 683–720; (b) Casu, B.; Lindahl, U. Adv. Carbohydr. Chem. Biochem. 2001, 57, 159–206; (c) Davis, B. G. Chem. Rev. 2002, 102, 579–601.
- 2. (a) Banoub, J.; Boullanger, P.; Lafont, D. Chem. Rev. 1992, 92, 1167–1195; (b) Debenham, J.; Rodebaugh, R.; Fraser-Reid, B. Liebigs Ann. Recl. 1997, 791–802.
- 3. Lemieux, R. U.; Ratcliffe, R. M. Can. J. Chem. 1979, 57, 1244–1251.
- 4. (a) Demchenko, A. V. Curr. Org. Chem. 2003, 7, 35–79; (b) Bongat, A. F. G.; Demchenko, A. V. Carbohydr. Res. 2007, 342, 374–406.
- 5. (a) Benakli, K.; Zha, C.; Kerns, R. J. J. Am. Chem. Soc. 2001, 123, 9461–9462; (b) Kerns, R. J.; Zha, C.; Benakli, K.; Liang, Y.-Z. Tetrahedron Lett. 2003, 44, 8069– 8072; (c) Wei, P.; Kerns, R. J. J. Org. Chem. 2005, 70, 4195–4198.
- 6. Boysen, M.; Gemma, E.; Lahmann, M.; Oscarson, S. Chem. Commun. 2005, 3044–3046.
- 7. Manabe, S.; Ishii, K.; Ito, Y. J. Am. Chem. Soc. 2006, 128, 10666–10667.
- 8. Geng, Y.; Zhang, L.-H.; Ye, X.-S. Chem. Commun. 2008, 597–599.
- 9. (a) Huang, X.; Huang, L.; Wang, H.; Ye, X.-S. Angew. Chem., Int. Ed. 2004, 43, $5221 - 5224$; (b) Huang, L.; Huang, X. Chem.—Eur. J. 2007, 13, 529–540; (c) Huang, L.; Wang, Z.; Li, X.; Ye, X.-S.; Huang, X. Carbohydr. Res. 2006, 341, 1669– 1679; (d) Wang, Z.; Zhou, L.; El-Boubbou, K.; Ye, X.-S.; Huang, X. J. Org. Chem. 2007, 72, 6409–6420; (e) Teumelsan, N.; Huang, X. J. Org. Chem. 2007, 72, 8976– 8979; (f) Wang, C.-C.; Lee, J.-C.; Luo, S.-Y.; Kulkarni, S. S.; Huang, Y.-W.; Lee, C.-C.; Chang, K.-L.; Hung, S.-C. Nature 2007, 446, 896–899.
- 10. (a) Wang, Y.; Ye, X.-S.; Zhang, L.-H. Org. Biomol. Chem. 2007, 5, 2189–2200; (b) Wang, Y.; Zhang, L.-H.; Ye, X.-S. Comb. Chem. High Throughput Screen. 2006, 9, 63–75; (c) Geng, Y.; Ye, X.-S. Prog. Chem. 2007, 19, 1896–1902.
- 11. (a) van der Plas, H. C.; Koudijs, A. Recl. Trav. Chim. Pays-Bas 1978, 97, 159–161; (b) Crich, D.; Smith, M.; Yao, Q.; Picione, J. Synthesis 2001, 323–326.
- 12. Zhang, Z.; Ollman, I. R.; Ye, X.-S.; Wischnat, R.; Baasov, T.; Wong, C.-H. J. Am. Chem. Soc. 1999, 121, 734–753.
- 13. Wang, C.; Wang, H.; Huang, X.; Zhang, L.-H.; Ye, X.-S. Synlett 2006, 2846–2850.
- 14. Donor 1d did not afford the expected disaccharide after it was absolutely activated by means of BSM/Tf_2O , presumably due to the instability of intermediate after the pre-activation. Donor 1f could be activated at a range of temperatures from -73 °C to room temperature by the use of promoter systemperatures from 12 and 12 and NIS/AgOTf
- 15. Crich and co-workers have convincingly demonstrated the presence of glycosyl triflate as the dominant intermediate after pre-activation of several thioglycosides through a series of low-temperature NMR experiments. For references regarding the reactive intermediate, see: (a) Crich, D.; Sun, S. J. Am. Chem. Soc. 1998, 120, 435–436; (b) Crich, D.; Sun, S. Tetrahedron 1998, 54, 8321–8348; (c) Crich, D.; Cai, W. J. Org. Chem. 1999, 64, 4926–4930.
- 16. Crich, D.; Smith, M. J. Am. Chem. Soc. 2001, 123, 9015–9020.
- 17. (a) Garcia, B. A.; Poole, J. L.; Gin, D. Y. J. Am. Chem. Soc. 1997, 119, 7597–7598; (b) Codee, J. D. C.; Litjens, R. E. J. N.; den Heeten, R.; Overkleeft, H. S.; van Boom, J. H.; van der Marel, G. A. Org. Lett. 2003, 5, 1519–1522.
- 18. Duron, S. G.; Polat, T.; Wong, C.-H. Org. Lett. 2004, 6, 839–841.
- 19. (a) Veenenman, G. H.; van Boom, J. H. Tetrahedron Lett. 1990, 31, 275–278; (b) Konradsson, P.; Udodong, U. E.; Fraser-Reid, B. Tetrahedron Lett. 1990, 31, 4313–4316.
- 20. In contrast to the basic condition of 1 M NaOH (see Ref. 7), this base condition afforded higher yield during a shorter period of time.